

# Sample Prep Water Soluble Fertilizers

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Lab Services

# Sampling Techniques and Small Sample Sizes can be an issue with Water Soluble Fertilizers that are made with Hygroscopic Components

- ▶ Results can be skewed which create violations, fines, and stop sales for fertilizers which do not meet their guarantee
- ▶ The total of the nutrients tested do not always show a balancing over and under
- ▶ Calcium and Magnesium levels are not always tested
- ▶ Larger Sampling Sizes and a different technique (scooping) works



Acid

Tank A  
5-12-26

Tank B  
Calcium  
Nitrate



Concentrate tanks 20- 50 gallons  
2-3 lb. dissolved per gallon  
Injector ratio 1:100 - 1:200

# Jack's LX 13-2-13

- ▶ Designed for plug production
- ▶ Over 93% nitrate nitrogen
- ▶ Ideal ratio of 6% calcium : 3% magnesium
- ▶ Strong cell wall improves post harvest performance
- ▶ Low P content to promote compact, sturdy plugs



# Jack's LX 13-2-13

- ▶ Key to success = high NO<sub>3</sub> and Ca content yet still can be tank mixed with K-Trate for a strong finish
- ▶ Reduces need to alternate with a different formula later in crop growth cycle

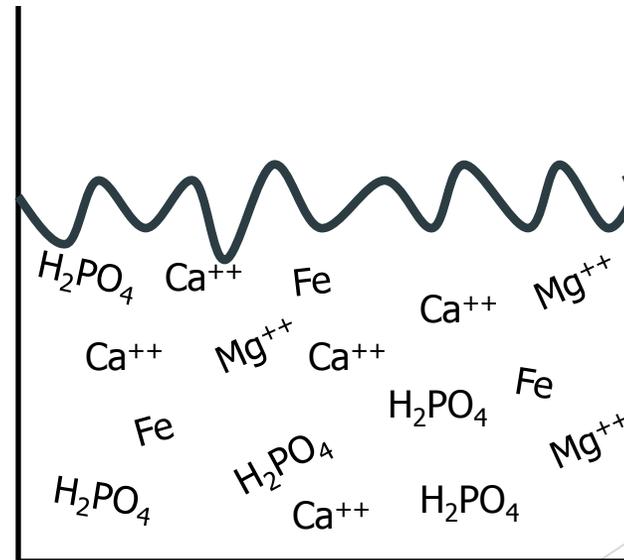
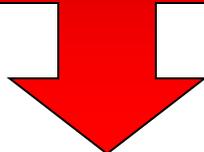


# Jack's LX Fertilizers

## Keeping Calcium & Magnesium in Solution

At low solution pH all nutrients can be mixed in one concentration tank without forming precipitates!

Solution  
pH  $\approx$  3-4



# 13-2-13 Plug LX Autopsy

Calcium Nitrate

Magnesium Nitrate

Ammonium Nitrate

MonoPotassium Phosphate

Potassium Nitrate

Trace Elements



# Raw Materials

Various SGN's and shapes

Very Hygroscopic

Cannot Grind

Difficult to Sample in small quantities



## ▶ J.R. Peters, Inc. - Water Soluble Fertilizer QA Method

▶ SOP#: JRP- 09

▶ Method Name: qafert

▶ Fertilizer Type: Professional Water Soluble Samples

### ▶ Scope:

▶ This procedure is for the preparation and analysis of samples of water-soluble professional fertilizer formulations. These water-soluble fertilizer (WSF) samples are designated by the internal lab login name of **qafert**. Samples will be analyzed on the ICP-AES Spectrometer for: P2O5%, K2O%, Ca%, Mg%, S%, Mn%, Mo%, B%, Fe%, Cu%, Zn%, Al%, and Na%, Sample will also be analyzed on the Skalar automated wet chemistry analyzer for: NO3-N%, NH4-N%, Urea-N%, and Cl%.

### ▶ Apparatus:

▶ Analytical Balance

▶ Volumetric Flasks

▶ Stir Plate and Magnetic Stir Bar

### **Batch Sample Procedure:**

1. For large batch samples weigh a **250g sub-sample** and place into a 1000ml volumetric flask.
2. Fill the flask 2/3 full with DI water.
3. Place mixture on stir plate. Insert stir bar and stir until visibly dissolved. Since this is an endothermic reaction, allow the sample to sit at room temperature for at least **15hrs** to ensure total sample dissolution and equilibration.
4. Remove the stir bar and bring up to volume with D.I. water. Mix thoroughly.
5. The resulting solution is a 25% fertilizer solution.
6. Make the necessary dilutions from this solution for the Skalar and ICP. (e.g. For a 1% solution use 20mls of the 25% solution in a 500ml flask. Bring to volume with DI water.)
7. Solution is analyzed using a 0.1% and 0.01% dilution for Skalar and in duplicate on ICP at 0.1% and 1% dilution as well as 0.01% dilution for high P or K containing samples.

# What can be done to modify the accepted procedure for sampling and testing these types of products to improve the results?

- ▶ Increase the number of core samples taken?
- ▶ Increase the size of the sample tested?
- ▶ Test present method versus modified method?
- ▶ Other suggestions